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## Nasal powder delivery – Characterisation of the influence of excipients on drug absorption

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# Nasal powder formulations fillers Introduction Influences on 1 Mucociliary clearance

(2) Dissolution

(3) Permeation

Easy access to a highly vascularised, relatively permeable mucosa via the nose offers high potential for systemic drug delivery. However, the natural mucociliary cleaning mechanism results in a short nasal residence time of drugs and thus poses a specific challenge. The formulation of nasal powders and the use of functional excipients are strategies to overcome this hurdle, but influence drug absorption multifactorially.



drug + mucoadhesives

permeation

enhancers

Therefore, the aim of this study was to distinguish influences of excipients on mucociliary clearance, dissolution and permeation, in order to enable a sound selection and thus the development of successful nasal products.

### **Characterisation methods** Dispersions of powders in Influence on simulated nasal fluid (SNF) mucociliary 7.45 g/L NaCl Frequency dependent 1.29 g/L KCl clearance using elastic and viscous moduli 0.32 g/L CaCl<sub>2</sub> x 2 H<sub>2</sub>O oscillation Displayed frequency: 1 Hz rheology Donor Influence on Membrane Powder on an air-liquid-2 drug dissolution interface Cellulose acetate (0.45 µm) using Franz Acceptor diffusion cells Temperature Simulated nasal fluid 32 °C **RPMI 2650** Influence on Test solultions Air-liquid-**TEER** drug permeation API + single interface $> 75 \Omega xcm^2$ using RPMI excipients in Transwell insert **2650 cells** HBSS + HEPES

## Assessed excipients and model formulations

PETP (3 µm)

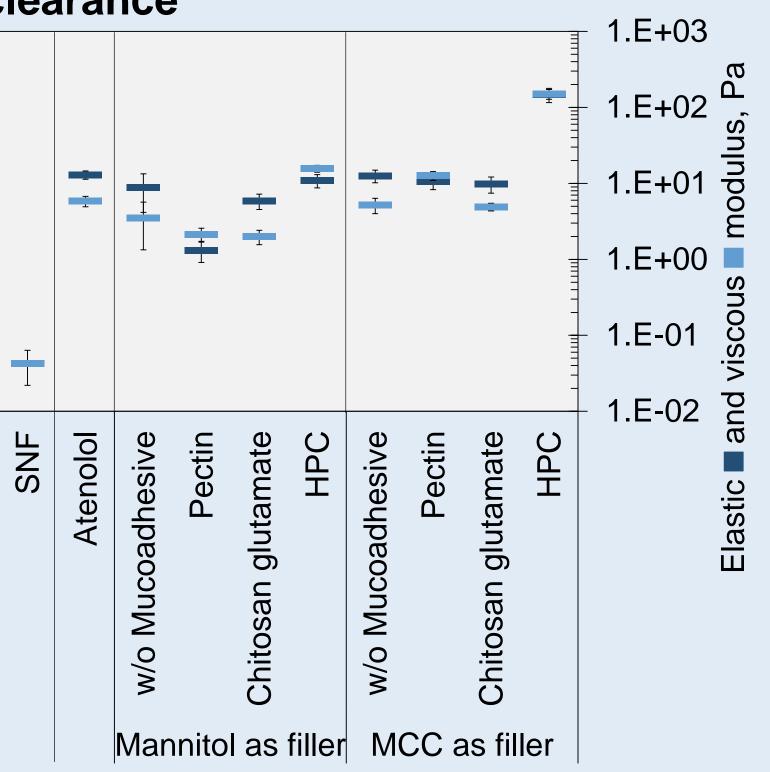
Function	Compound	Characteristics	Formulations		
			Control	w/o MA	With MA
API	Atenolol	Low and paracellular permeation marker	100%	40%	40%
Fillers	Mannitol	Water-soluble	_	60%	50%
	Microcrystalline cellulose (MCC)	Water-insoluble			
Muco- adhesives (MA)	Pectin	Anionic charge	_		
	Chitosan glutamate	Salt of positively charged chitosan			10%
	Hydroxypropyl cellulose (HPC)	Neutral	_		

The excipients were fractioned on a laboratory sieve shaker and a sieve fraction of 32-150 µm was used for the experiments. Model formulations were prepared by blending API and excipients in a Turbula blender.

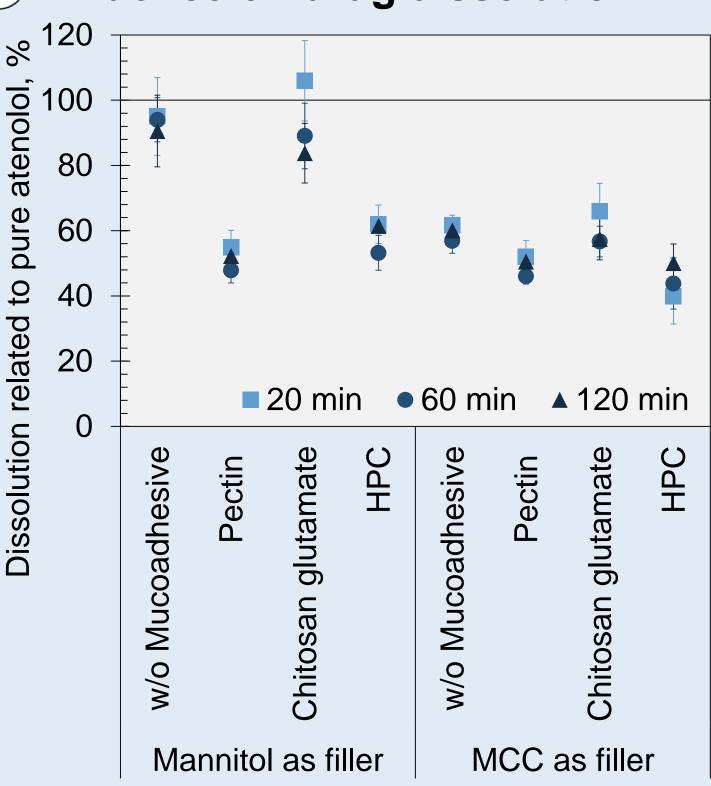
### Results and discussion

1 Influence on mucociliary clearance

Undissolved particles drug increased the elasticity of SNF sufficiently mucociliary clearance (>2 Pa). However, effect decrease with progressing drug dissolution in the nose. The use insoluble or swelling excipients can induce a longer lasting effect. An additional increase in elastic and viscous modulus compared to pure API was only observed with the formulation containing MCC and HPC.

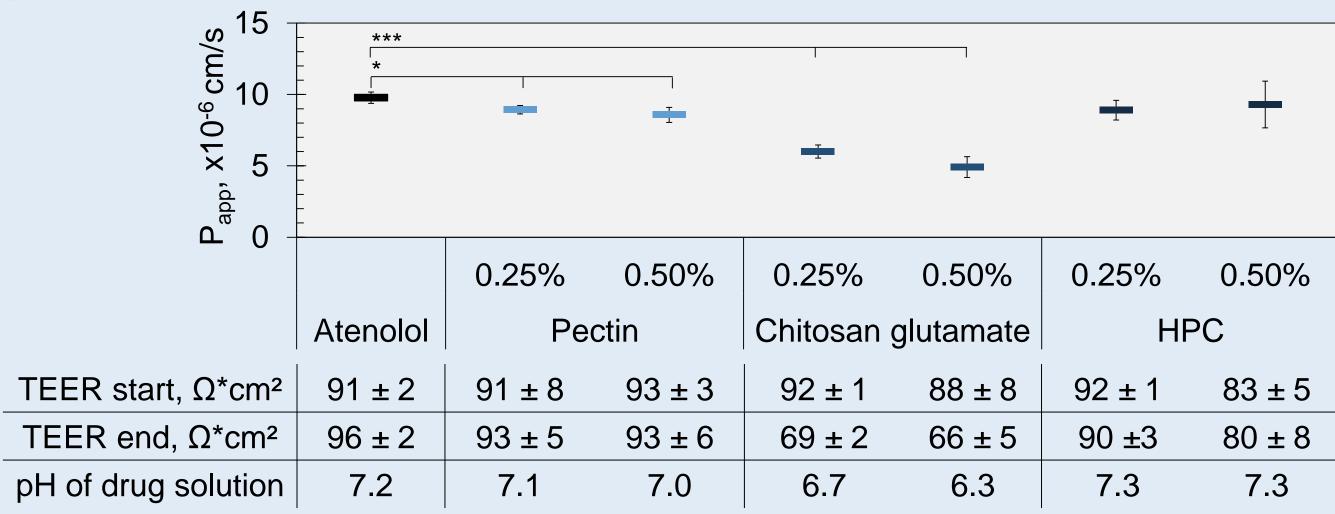


### 2 Influence on drug dissolution



Formulations containing mannitol with or w/o chitosan glutamate showed a similar dissolution profile as the pure API. All other formulations, containing insoluble or gelling excipients, decreased the dissolution rate similarly. Considering the results of the rheological assessments (1), the formulation containing MCC and HPC is expected to show the greatest benefit if a prolonged nasal residence time is required for drug permeation.

### 3 Influence on drug permeation



Pectin and chitosan glutamate reduced the permeability of atenolol. Chitosan glutamate however, reduced the TEER of the cell barrier, which supports the postulated function of chitosan to enhance permeation by opening tight junctions. The slight acidic reaction of chitosan glutamate may contribute to the yet reduced permeability. A higher portion of atenolol is in its ionised form at lower pH, which contributes less to permeation. The acidic reaction of chitosan glutamate may limit its use with drugs that are weak bases, as pH adjustment is difficult in powder formulations.

Mean +/- standard deviation (n=3; pH: n=1 \*=p<0.05; \*\*\*=p<0.001)

### Conclusion



The use of excipients in nasal powder formulations can specifically adjust drug absorption. The separate characterisation of processes that influence drug absorption is essential for a sound selection of functional excipients. The methods described in this work enable the detection of counteracting effects on absorption and thus, the selection of excipients with the most promising overall properties.

### Acknowledgements

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